

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
29 January 2004 (29.01.2004)

PCT

(10) International Publication Number
WO 2004/009699 A1

(51) International Patent Classification⁷: C08L 23/16,
C10G 2/00, 45/58

(21) International Application Number:
PCT/EP2003/008063

(22) International Filing Date: 16 July 2003 (16.07.2003)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
02016087.5 19 July 2002 (19.07.2002) EP

(71) Applicant (for all designated States except US): SHELL
INTERNATIONALE RESEARCH MAATSCHAPPIJ
B.V. [NL/NL]; Carel van Bylandtlaan 30, NL-2596 HR
The Hague (NL).

(72) Inventor; and

(75) Inventor/Applicant (for US only): NULL, Volker, Klaus
[DE/DE]; Hohe-Schaar-Strasse 36, 21107 Hamburg (DE).

(81) Designated States (national): AE, AG, AL, AM, AT, AU,
AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU,
CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,
MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC,
SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA,
UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (regional): ARIPO patent (GH, GM,
KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW),
Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM),
European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE,
ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO,
SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM,
GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guid-
ance Notes on Codes and Abbreviations" appearing at the begin-
ning of each regular issue of the PCT Gazette.



WO 2004/009699 A1

(54) Title: COMPOSITION COMPRISING EPDM AND A PARAFFINIC OIL

(57) Abstract: Composition comprising a ethylene-propylene-diene rubber component and a Fischer-Tropsch derived process oil.
The process oil preferably has a flash point of above 260 °C and an evaporation loss at 107 °C during 22 hours of less than 0.05 wt%.

COMPOSITION COMPRISING EPDM AND A PARAFFINIC OIL

The invention is directed to a composition comprising an ethylene-propylene-diene (EPDM) rubber component and a paraffinic oil component.

Such a EPDM containing compositions are well known
5 and are described in detail in Rubber Technology Handbook, Werner Hofmann, Oxford University Press, New York, 1980, paragraph 3.3.8, pages 93-100. Paraffinic oils are used as plasticisers or softening agent in such compositions. Also compositions often
10 referred to as thermoplastic elastomers (TPE) or also sometimes referred to as thermoplastic vulcanisates (TPV) as described in said Handbook on pages 144 - 150 and 154 - 156 comprising EPDM and a polypropylene may comprise paraffinic process oils. Examples of such
15 compositions may be found in EP-A-1132242, GB-A-155020, US-A-5290886 and EP-A-1028145.

The paraffinic oils commonly used in such applications are either hydroprocessed petroleum derived oils, such as the Paralux oils as supplied by Chevron
20 Products Company or synthetic paraffin oils, for example poly-alpha olefins such as a low weight component of Poly- ethylene-propylene such as described in EP-A-1028145.

US-A-4208310 and US-A-4134870 describe an elastomer
25 composition containing a paraffinic wax, which according to the description may be a Fischer-Tropsch derived wax. The addition of the wax, which is solid at room temperature, is improved physical and rheological properties.

EP-A-577255 describes an EPDM composition which contain an extender oil and a crystalline paraffin wax. The crystalline wax may be prepared by a Fischer-Tropsch process. The paraffinic wax has a melting point of
5 between 60 and 100 °C and is thus solid at room temperature.

Compositions as described above are often used in automotive applications, such as parts of the interior of the automobile. There is an increasing demand for low
10 hydrocarbon emissions of an automobile. These hydrocarbon emissions are measured by keeping a complete car at an elevated temperature and detecting any hydrocarbon emissions. In view of this development there is an increasing demand for EPDM containing compositions having
15 a low hydrocarbon emission.

The object of this invention is to provide an EPDM containing composition having a low hydrocarbon emission.

This object is achieved by the following composition. A composition comprising an ethylene-propylene-diene
20 (EPDM) rubber component and a paraffinic oil component, wherein the paraffinic oil component is a Fischer-Tropsch derived process oil.

Applicants have found that a process oil as derived from a Fischer-Tropsch synthesis product can be simply
25 obtained having properties which lower the hydrocarbon emissions of the finished EPDM comprising product. Some severely hydroprocessed or synthetic paraffinic process oils as described above may also achieve this lower hydrocarbon emission. A disadvantage of these
30 products is that they are either very expensive because they have to be synthesized from lower olefins or by heavily hydroprocessing. Another advantage of the Fischer-Tropsch derived oils as compared to the heavily

hydroprocessed oils is that the low temperature properties for the higher viscosity grade oils is much better making the Fischer-Tropsch derived oils more easy to handle in the process to make the EDPM containing product.

The Fischer-Tropsch derived oil preferably has a flash point according to ISO 2592 of above 240 °C and more preferably above 260 °C. The UV adsorption of the oil at 300 nm is preferably less than 1% and more preferably less than 0.6% according to ASTM D 2008-A1. The evaporation loss at 107 °C during 22 hours of the oil (according to ASTM D 972 is preferably less than 0.1 wt% and more preferably less than 0.05 wt%.

The kinematic viscosity at 100 °C of the oil will be resultant from the above requirements and will usually be above 8 cSt and more preferably above 9 cSt. The upper limit is not essential for the hydrocarbon emissions requirements. Nevertheless it is not advantageous to use too viscous oil for practical processing reasons.

Preferably the upper limit will be around 30 cSt, preferably 25 cSt. The pour point of the process oil will be dependent on the viscosity grade used. Applicants have found a process involving especially a catalytic dewaxing step to prepare a Fischer-Tropsch process oil having pour points ranging from -40 °C for the less viscous grades to around 10 °C for the more viscous grades. This has been found very advantageous because the oil can be used in a liquid state at ambient conditions during the manufacturing of the EPDM comprising composition.

Applicants further found that the Fischer-Tropsch derived oil preferably has a CN number as measured according to IEC 590 of between 15 and 30%. The oil is preferably liquid at 20 °C.

The process oil is preferably prepared using the below process, by

(a) hydrocracking/hydroisomerisating a feed comprising a Fischer-Tropsch derived feed,

5 (b) isolating from the product of step (a) a process oil precursor fraction,

(c) dewaxing the process oil precursor fraction obtained in step (b) to obtain the process oil, optionally after separating a lower boiling fraction from said dewaxed
10 product.

The Fischer-Tropsch derived feed can be obtained by well-known processes, for example the so-called commercial Sasol process, the commercial Shell Middle Distillate Process or by the non-commercial
15 Exxon process. These and other processes are for example described in more detail in EP-A-776959, EP-A-668342, US-A-4943672, US-A-5059299, WO-A-9934917 and WO-A-9920720.

A preferred process to prepare the process oil having
20 the desired flash point, evaporation loss and CN-number properties involves using a Fischer-Tropsch derived feed in step (a) which is characterized in that the weight ratio of compounds having at least 60 or more carbon atoms and compounds having at least 30 carbon atoms in
25 the Fischer-Tropsch derived feed is at least 0.2 and wherein at least 30 wt% of compounds in the Fischer-Tropsch product have at least 30 carbon atoms. More preferably the feed has at least 50 wt% and most preferably at least 55 wt% of compounds having at
30 least 30 carbon atoms. Furthermore the weight ratio of compounds having at least 60 or more carbon atoms and compounds having at least 30 carbon atoms of the Fischer-Tropsch derived feed is preferably at

least 0.4 and more preferably at least 0.55. The Fischer-Tropsch derived feed is preferably derived from a Fischer-Tropsch product which comprises a C₂₀+ fraction having an ASF-alpha value (Anderson-Schulz-Flory chain growth factor) of at least 0.925, preferably at
5 least 0.935, more preferably at least 0.945, even more preferably at least 0.955.

The initial boiling point of the Fischer-Tropsch derived feed may range up to 400 °C, but is preferably
10 below 200 °C. Preferably at least any compounds having 4 or less carbon atoms and any compounds having a boiling point in that range are separated from a Fischer-Tropsch synthesis product before the Fischer-Tropsch synthesis product is used as a
15 Fischer-Tropsch derived feed in step (a). In addition to this Fischer-Tropsch product also other fractions may be part of the Fischer-Tropsch derived feed. Possible other fractions may suitably be any high boiling fraction obtained in step (b).

20 Such a Fischer-Tropsch derived feed is suitably obtained by a Fischer-Tropsch process, which yields a relatively heavy Fischer-Tropsch product. Not all Fischer-Tropsch processes yield such a heavy product. An example of a suitable Fischer-Tropsch process is
25 described in WO-A-9934917 and in AU-A-698392. These processes may yield a Fischer-Tropsch product as described above.

The Fischer-Tropsch derived feed and the resulting waxy raffinate product will contain no or very little
30 sulphur and nitrogen containing compounds. This is typical for a product derived from a Fischer-Tropsch reaction, which uses synthesis gas containing almost no impurities. Sulphur and nitrogen levels will generally be

below the detection limits, which are currently 5 ppm for sulphur and 1 ppm for nitrogen

The Fischer-Tropsch product may optionally be subjected to a mild hydrotreatment step in order to remove any oxygenates and saturate any olefinic compounds present in the reaction product of the Fischer-Tropsch reaction. Such a hydrotreatment is described in EP-B-668342. The mildness of the hydrotreating step is preferably expressed in that the degree of conversion in this step is less than 20 wt% and more preferably less than 10 wt%. The conversion is here defined as the weight percentage of the feed boiling above 370 °C that reacts to a fraction boiling below 370 °C. After such a mild hydrotreatment lower boiling compounds, having three or less carbon atoms and other compounds boiling in that range, will preferably be removed from the effluent before it is used in step (a).

The hydrocracking/hydroisomerisation reaction of step (a) is preferably performed in the presence of hydrogen and a catalyst, which catalyst can be chosen from those known to one skilled in the art as being suitable for this reaction. Catalysts for use in step (a) typically comprise an acidic functionality and a hydrogenation/dehydrogenation functionality. Preferred acidic functionalities are refractory metal oxide carriers. Suitable carrier materials include silica, alumina, silica-alumina, zirconia, titania and mixtures thereof. Preferred carrier materials for inclusion in the catalyst for use in the process of this invention are silica, alumina and silica-alumina. A particularly preferred catalyst comprises platinum supported on a silica-alumina carrier. If desired, the acidity of the catalyst carrier may be enhanced by applying a halogen

moiety, in particular fluorine, or a phosphorous moiety to the carrier. Examples of suitable hydrocracking/hydroisomerisation processes and suitable catalysts are described in WO-A-0014179, EP-A-532118 and the earlier referred to EP-A-776959.

Preferred hydrogenation/dehydrogenation functionalities are Group VIII metals, such as nickel, cobalt, iron, palladium and platinum. Preferred are the noble metal Group VIII members, palladium and more preferred platinum. The catalyst may comprise the more preferred noble metal hydrogenation/dehydrogenation active component in an amount of from 0.005 to 5 parts by weight, preferably from 0.02 to 2 parts by weight, per 100 parts by weight of carrier material. A particularly preferred catalyst for use in the hydroconversion stage comprises platinum in an amount in the range of from 0.05 to 2 parts by weight, more preferably from 0.1 to 1 parts by weight, per 100 parts by weight of carrier material. The catalyst may also comprise a binder to enhance the strength of the catalyst. The binder can be non-acidic. Examples are clays and other binders known to one skilled in the art.

In step (a) the feed is contacted with hydrogen in the presence of the catalyst at elevated temperature and pressure. The temperatures typically will be in the range of from 175 to 380 °C, preferably higher than 250 °C and more preferably from 300 to 370 °C. The pressure will typically be in the range of from 10 to 250 bar and preferably between 20 and 80 bar. Hydrogen may be supplied at a gas hourly space velocity of from 100 to 10000 NL/l/hr, preferably from 500 to 5000 NL/l/hr. The hydrocarbon feed may be provided at a weight hourly space velocity of from 0.1 to 5 kg/l/hr, preferably higher than

0.5 kg/l/hr and more preferably lower than 2 kg/l/hr. The ratio of hydrogen to hydrocarbon feed may range from 100 to 5000 Nl/kg and is preferably from 250 to 2500 Nl/kg.

5 The conversion in step (a) as defined as the weight percentage of the feed boiling above 370 °C which reacts per pass to a fraction boiling below 370 °C, is at least 20 wt%, preferably at least 25 wt%, but preferably not more than 80 wt%, more preferably not more than 65 wt%.

10 The feed as used above in the definition is the total hydrocarbon feed fed to step (a), thus also any optional recycle of the higher boiling fraction as obtained in step (b).

15 In step (b) the product of step (a) is preferably separated into one or more distillate fractions, a process oil precursor fraction, preferably having a T10wt% boiling point of between 300 and 450 °C. A heavy fraction may be separated from the product of step (a) to adjust the resultant viscosity of the process oil. If no

20 heavy fraction is removed the kinematic viscosity at 100 °C of the process oil may be well above 15 cSt. By adjusting the amount and cut point at which the said heavy fraction is separated from the effluent of step (a) process oils can be obtained having a kinematic viscosity

25 at 100 °C ranging from 8 cSt cSt to above 25 cSt.

 In step (c) the process oil precursor fraction obtained in step (b) is subjected to a dewaxing treatment wherein the pour point of the oil is reduced. The pour point is preferably reduced by more than 10 °C.

30 Dewaxing can be performed by means of a so-called solvent dewaxing process or by means of a catalytic dewaxing process.

Solvent dewaxing is well known to those skilled in the art and involves admixture of one or more solvents and/or wax precipitating agents with the process oil precursor fraction and cooling the mixture to a temperature in the range of from -10 °C to -40 °C, preferably in the range of from -20 °C to -35 °C, to separate the wax from the oil. The oil containing the wax is usually filtered through a filter cloth which can be made of textile fibres, such as cotton; porous metal cloth; or cloth made of synthetic materials. Examples of solvents which may be employed in the solvent dewaxing process are C₃-C₆ ketones (e.g. methyl ethyl ketone, methyl isobutyl ketone and mixtures thereof), C₆-C₁₀ aromatic hydrocarbons (e.g. toluene), mixtures of ketones and aromatics (e.g. methyl ethyl ketone and toluene), autorefrigerative solvents such as liquefied, normally gaseous C₂-C₄ hydrocarbons such as propane, propylene, butane, butylene and mixtures thereof. Mixtures of methyl ethyl ketone and toluene or methyl ethyl ketone and methyl isobutyl ketone are generally preferred. Examples of these and other suitable solvent dewaxing processes are described in Lubricant Base Oil and Wax Processing, Avilino Sequeira, Jr, Marcel Dekker Inc., New York, 1994, Chapter 7.

Preferably step (c) is performed by means of a catalytic dewaxing process. The catalytic dewaxing process can be performed by any process wherein in the presence of a catalyst and hydrogen the pour point of the process oil precursor fraction is reduced as specified above. Suitable dewaxing catalysts are heterogeneous catalysts comprising a molecular sieve and optionally in combination with a metal having a hydrogenation function, such as the Group VIII metals. Molecular sieves, and more

suitably intermediate pore size zeolites, have shown a good catalytic ability to reduce the pour point of the process oil precursor fraction under catalytic dewaxing conditions. Preferably the intermediate pore size zeolites have a pore diameter of between 0.35 and 0.8 nm. Suitable intermediate pore size zeolites are mordenite, ZSM-5, ZSM-12, ZSM-22, ZSM-23, SSZ-32, ZSM-35 and ZSM-48. Another preferred group of molecular sieves are the silica-aluminaphosphate (SAPO) materials of which SAPO-11 is most preferred as for example described in US-A-4859311. ZSM-5 may optionally be used in its HZSM-5 form in the absence of any Group VIII metal. The other molecular sieves are preferably used in combination with an added Group VIII metal. Suitable Group VIII metals are nickel, cobalt, platinum and palladium. Examples of possible combinations are Ni/ZSM-5, Pt/ZSM-23, Pd/ZSM-23, Pt/ZSM-48 and Pt/SAPO-11. Further details and examples of suitable molecular sieves and dewaxing conditions are for example described in WO-A-9718278, US-A-5053373, US-A-5252527 and US-A-4574043.

The dewaxing catalyst suitably also comprises a binder. The binder can be a synthetic or naturally occurring (inorganic) substance, for example clay, silica and/or metal oxides. Natural occurring clays are for example of the montmorillonite and kaolin families. The binder is preferably a porous binder material, for example a refractory oxide of which examples are: alumina, silica-alumina, silica-magnesia, silica-zirconia, silica-thoria, silica-beryllia, silica-titania as well as ternary compositions for example silica-alumina-thoria, silica-alumina-zirconia, silica-alumina-magnesia and silica-magnesia-zirconia. More preferably a low acidity refractory oxide binder

material, which is essentially free of alumina, is used. Examples of these binder materials are silica, zirconia, titanium dioxide, germanium-dioxide, boria and mixtures of two or more of these of which examples are listed above. The most preferred binder is silica.

A preferred class of dewaxing catalysts comprise intermediate zeolite crystallites as described above and a low acidity refractory oxide binder material which is essentially free of alumina as described above, wherein the surface of the aluminosilicate zeolite crystallites has been modified by subjecting the aluminosilicate zeolite crystallites to a surface dealumination treatment. A preferred dealumination treatment is by contacting an extrudate of the binder and the zeolite with an aqueous solution of a fluorosilicate salt as described in for example US-A-5157191 or WO-A-0029511. Examples of suitable dewaxing catalysts as described above are silica bound and dealuminated Pt/ZSM-5, silica bound and dealuminated Pt/ZSM-23, silica bound and dealuminated Pt/ZSM-12, silica bound and dealuminated Pt/ZSM-22, as for example described in WO-A-0029511 and EP-B-832171.

Catalytic dewaxing conditions are known in the art and typically involve operating temperatures in the range of from 200 to 500 °C, suitably from 250 to 400 °C, hydrogen pressures in the range of from 10 to 200 bar, preferably from 40 to 70 bar, weight hourly space velocities (WHSV) in the range of from 0.1 to 10 kg of oil per litre of catalyst per hour (kg/l/hr), suitably from 0.2 to 5 kg/l/hr, more suitably from 0.5 to 3 kg/l/hr and hydrogen to oil ratios in the range of from 100 to 2,000 litres of hydrogen per litre of oil. By varying the temperature between 315 and 375 °C at

between 40-70 bars, in the catalytic dewaxing step it is possible to prepare base oils having different pour point specifications varying from suitably -10 to -60 °C.

5 Optionally a lower boiling fraction is separated from the oil as obtained after dewaxing. The need to separate a fraction will be determined by the properties of the process oil precursor fraction used in step (c) and the dewaxing process used. For example a catalytic dewaxing step will suitably require such a separation because
10 lower boiling fractions are formed in the dewaxing process, which need to be removed in order to achieve the volatility requirements of the process oil used in the present invention.

15 The process oil may be subjected to a hydrofinishing step or an adsorption step in order to improve its colour properties. Adsorption may be performed by contacting the oil with a suitable heterogeneous adsorbents, for example active carbon, zeolites, for example natural faujasite, or synthetic materials such as ferrierite, ZSM-5,
20 faujasite, mordenite, metal oxides such as silica powder, silica gel, aluminium oxide and various clays, for example Attapulugus clay (hydrous magnesium-aluminium silicate), Porocel clay (hydrated aluminium oxide). A preferred adsorbent is activated carbon.

25 A hydrofinishing step is suitably carried out at a temperature between 180 and 380 °C, a total pressure of between 10 to 250 bar and preferably above 100 bar and more preferably between 120 and 250 bar. The WHSV (Weight hourly space velocity) ranges from 0.3 to 2 kg of oil per
30 litre of catalyst per hour (kg/l.h).

 The hydrogenation catalyst is suitably a supported catalyst comprising a dispersed Group VIII metal. Possible Group VIII metals are cobalt, nickel, palladium

and platinum. Cobalt and nickel containing catalysts may also comprise a Group VIB metal, suitably molybdenum and tungsten. Suitable carrier or support materials are amorphous refractory oxides. Examples of suitable amorphous refractory oxides include inorganic oxides, such as alumina, silica, titania, zirconia, boria, silica-alumina, fluorided alumina, fluorided silica-alumina and mixtures of two or more of these.

Examples of suitable hydrogenation catalysts are nickel-molybdenum containing catalyst such as KF-847 and KF-8010 (AKZO Nobel) M-8-24 and M-8-25 (BASF), and C-424, DN-190, HDS-3 and HDS-4 (Criterion); nickel-tungsten containing catalysts such as NI-4342 and NI-4352 (Engelhard) and C-454 (Criterion); cobalt-molybdenum containing catalysts such as KF-330 (AKZO-Nobel), HDS-22 (Criterion) and HPC-601 (Engelhard). Preferably platinum containing and more preferably platinum and palladium containing catalysts are used. Preferred supports for these palladium and/or platinum containing catalysts are amorphous silica-alumina. Examples of suitable silica-alumina carriers are disclosed in WO-A-9410263. A preferred catalyst comprises an alloy of palladium and platinum preferably supported on an amorphous silica-alumina carrier of which the commercially available catalyst C-624 and C-654 of Criterion Catalyst Company (Houston, TX) are examples.

The content of the Fischer-Tropsch derived paraffinic process oil in the composition according to the present invention will depend on the desired properties of the end product and on the other components of the composition. The oil is usually applied as plasticiser. Typical ranges are mentioned in the above referred to patent applications. Typically the content of the

paraffinic process oil may be between 1 and 60 wt% of the composition.

The composition may have the same composition as a commercial EPDM rubber wherein as the paraffinic oil the Fischer-Tropsch derived process oil is present.

Examples of commercial EPDM rubbers are the KELTAN EPDM rubbers from DSM Elastomers, the VISTALON EPDM rubbers from ExxonMobil Chemical and DUTRAL EPDM rubbers from Enichem. (KELTAN, VISTALON, DUTRAL are trademarks)

The composition may also be a thermoplastic vulcanisate composition comprising an ethylene-propylene-diene rubber component and a poly-olefin component. The poly-alpha olefin is preferably polypropylene. Such compositions combines the elastic properties of a rubber with the processing characteristics of a thermoplast. The composition comprises preferably EPDM particles embedded in a polypropylene (PP) matrix. The PP phase presents the processing behaviour of a PP, while the cured EPDM rubber provides excellent elastic properties. The composition may have the same composition as a commercial TPE composition, wherein as the paraffinic oil the Fischer-Tropsch derived process oil is present. Example of commercial thermoplastic vulcanisate are the KELTAN EPDM or SARLINK TPV series from DSM Elastomers (SARLINK is a trademark).

The below example will illustrate the preparation of a Fischer-Tropsch process oil having the desired properties for use in a composition according to the invention.

Example 1

The C₅-C₇₅₀ °C⁺ fraction of the Fischer-Tropsch product, as obtained in Example VII using the catalyst

of Example III of WO-A-9934917, was continuously fed to a hydrocracking step (step (a)). The feed contained about 60 wt% C₃₀+ product. The ratio C₆₀+ / C₃₀+ was about 0.55. In the hydrocracking step the fraction was contacted with a hydrocracking catalyst of Example 1 of EP-A-532118. The effluent of step (a) was continuously distilled under vacuum to give lights, fuels and a residue "R" boiling from 370 °C and above. The yield of gas oil fraction on fresh feed to hydrocracking step was 43 wt%. The properties of the gas oil thus obtained are presented in Table 3.

The main part of the residue "R" was recycled to step (a) and a remaining part was sent to a catalytic dewaxing step (c). The conditions in the hydrocracking step (a) were: a fresh feed Weight Hourly Space Velocity (WHSV) of 0.8 kg/l.h, recycle feed WHSV of 0.25 kg/l.h, hydrogen gas rate = 1000 Nl/kg, total pressure = 40 bar, and a reactor temperature of 335 °C.

In the dewaxing step, the fraction described above boiling from 370 °C to above 750 °C was contacted with a dealuminated silica bound ZSM-5 catalyst comprising 0.7% by weight Pt and 30 wt% ZSM-5 as described in Example 9 of WO-A-0029511. The dewaxing conditions were 40 bar hydrogen, WHSV = 1 kg/l.h and a temperature of 365 °C.

The dewaxed oil was distilled into a process oil fraction boiling above 510 °C and a fraction boiling below said fraction. The yield of process oil based on feed to dewaxing step was 27.9 wt%. The process oil (Process oil-1) was analyzed in more detail and the properties are listed in Table 1.

Example 2

5 A Shell MDS Waxy raffinate was contacted with a dealuminated silica bound ZSM-5 catalyst comprising 0.7% by weight Pt and 30 wt% ZSM-5 as described in Example 9 of WO-A-0029511. The dewaxing conditions were 40 bar hydrogen, WHSV = 1 kg/1.h and a temperature of 345 °C.

10 The dewaxed oil was distilled into a process oil fraction (Process oil 2) boiling having an initial boiling point as 466 °C and a 90 wt% boiling point of 567 °C. The yield of process oil on waxy raffinate feed was 12.3 wt%. Other properties of the process oil are listed in Table 1.

Table 1

	Process oil-1	Process oil-2
density at 20 °C	837.0	831.5
pour point (°C)	+9	-39
kinematic viscosity at 40 °C (cSt)		56.6
kinematic viscosity at 100 °C (cSt)	22.92	9.1
VI	178	139
sulphur content (%w)	< 0.001	< 0.001
Flash point (°C) (ISO 2592)	> 300	276
UV absorption at 300 nm (ASTM D 2008-A1)	< 0.6	< 0.6
Evaporation loss at 107 °C after 22 hours (ASTM D 972)	< 0.05	< 0.05
CN number (IEC 590)	Not measure d	18.6

C L A I M S

1. Composition comprising a ethylene-propylene-diene rubber component and a Fischer-Tropsch derived process oil.
2. Composition according to claim 1, wherein the
5 Fischer-Tropsch derived oil has a flash point of above 260 °C.
3. Composition according to any one of claims 1-2, wherein the UV adsorption of the process oil at 300 nm is less than 0.6% according to ASTM D 2008-A1.
- 10 4. Composition according to any one of claims 1-3, wherein the evaporation loss of the process oil at 107 °C during 22 hours is less than 0.05 wt%.
5. Composition according to any one of claims 1-4, wherein the kinematic viscosity at 100 °C of the process
15 oil is greater than 8 cSt.
6. Composition according to any one of claims 1-5, wherein the pour point of the process oil is below 10 °C.
7. Composition according to claim 6, wherein the process oil has a pour point of below 10 °C and a kinematic
20 viscosity at 100 °C of above 9 cSt.
8. Composition according to any one of claims 1-7, wherein the process oil is obtained by
 - (a) hydrocracking/hydroisomerisating a feed comprising a Fischer-Tropsch synthesis product,
 - 25 (b) isolating from the product of step (a) a process oil precursor fraction,
 - (c) dewaxing the process oil precursor fraction obtained in step (b) to obtain the process oil, optionally after

separating a lower boiling fraction from said dewaxed product.

9. Composition according to claim 8, wherein step (c) is performed by solvent dewaxing.

5 10. Composition according to claim 8, wherein step (c) is performed by catalytic dewaxing.

11. Composition according to any one of claims 8-10, wherein the conversion in step (a) is between 25 and 70 wt%.

10 12. Thermoplastic vulcanisate composition according to any one of claims 1-11, wherein the composition comprises a Ethylene-propylene-diene rubber component and a poly-olefin component.

15 13. Thermoplastic vulcanisate composition according to claim 12, wherein the poly-olefin is polypropylene.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 03/08063

A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C08L23/16 C10G2/00 C10G45/58		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC 7 C08L C10G C07C		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practical, search terms used) EPO-Internal, PAJ, WPI Data		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 134 870 A (MAKOWSKI HENRY S ET AL) 16 January 1979 (1979-01-16) abstract; claims 1-36 column 9, line 8-10; table 3 ---	1,5
X	US 4 208 310 A (BOCK JAN ET AL) 17 June 1980 (1980-06-17) abstract; claims 1-35 column 7; tables 1,2 column 9, line 7-40 ---	1,5,6
X	EP 0 577 255 A (SUMITOMO CHEMICAL CO) 5 January 1994 (1994-01-05) abstract; claims 1-11 page 2, line 49-58 page 3, line 7-12,22-25 ---	1,12,13
Y	abstract; claims 1-11 page 2, line 49-58 page 3, line 7-12,22-25 ---	1-13
-/-		
<input checked="" type="checkbox"/> Further documents are listed in the continuation of box C. <input checked="" type="checkbox"/> Patent-family members are listed in annex.		
* Special categories of cited documents : *A* document defining the general state of the art which is not considered to be of particular relevance *E* earlier document but published on or after the international filing date *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) *O* document referring to an oral disclosure, use, exhibition or other means *P* document published prior to the international filing date but later than the priority date claimed *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. *Z* document member of the same patent family		
Date of the actual completion of the international search 17 October 2003		Date of mailing of the international search report 29/10/2003
Name and mailing address of the ISA European Patent Office, P.O. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx 31 651 epo nl, Fax: (+31-70) 340-3016		Authorized officer Bergmans, K

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 03/08063

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 6 332 974 B1 (RYAN DANIEL F ET AL) 25 December 2001 (2001-12-25) abstract; claims 1-22 column 1, line 5-9 column 2, line 25-30 column 11; table 2	1-13
Y	US 3 102 098 A (MATUSZAK ALFRED H ET AL) 27 August 1963 (1963-08-27) abstract; claims 1-9 column 1, line 10-15 column 2, line 4, 55-62 column 4; table 1	1-13
A	WO 01 46339 A (CHEVRON USA INC) 28 June 2001 (2001-06-28) abstract; claims 1-34	1-13
A	EP 0 776 959 A (SHELL INT RESEARCH) 4 June 1997 (1997-06-04) cited in the application abstract; claims 1-10	1-13

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 03/08063

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4134870	A	16-01-1979	NONE
US 4208310	A	17-06-1980	DE 2861488 D1 18-02-1982 EP 0002572 A1 27-06-1979 IT 1101627 B 07-10-1985 JP 54090250 A 17-07-1979
EP 0577255	A	05-01-1994	JP 2792344 B2 03-09-1998 JP 6001888 A 11-01-1994 CA 2096218 A1 19-12-1993 DE 69315386 D1 08-01-1998 DE 69315386 T2 09-04-1998 EP 0577255 A2 05-01-1994
US 6332974	B1	25-12-2001	AU 750548 B2 18-07-2002 AU 5680899 A 03-04-2000 BR 9913583 A 22-05-2001 CA 2341607 A1 23-03-2000 EP 1144551 A2 17-10-2001 JP 2002538232 T 12-11-2002 NO 20011245 A 10-05-2001 TW 495548 B 21-07-2002 WO 0015736 A2 23-03-2000 ZA 200101684 A 28-05-2002
US 3102098	A	27-08-1963	NONE
WO 0146339	A	28-06-2001	US 6562230 B1 13-05-2003 AU 2431901 A 03-07-2001 CN 1425052 T 18-06-2003 EP 1244759 A2 02-10-2002 WO 0146339 A2 28-06-2001
EP 0776959	A	04-06-1997	EP 0776959 A2 04-06-1997 AU 705415 B2 20-05-1999 AU 7198896 A 05-06-1997 CA 2191290 A1 29-05-1997 CN 1167811 A ,B 17-12-1997 JP 9221685 A 26-08-1997 SG 74001 A1 18-07-2000 TW 416981 B 01-01-2001 ZA 9609906 A 28-05-1997